



General Certificate of Secondary Education

*Additional Science 4463 /
Physics 4451*

PHY2F Unit Physics 2

Mark Scheme

2012 Examination – January Series

Mark schemes are prepared by the Principal Examiner and considered, together with the relevant questions, by a panel of subject teachers. This mark scheme includes any amendments made at the standardisation meeting attended by all examiners and is the scheme which was used by them in this examination. The standardisation meeting ensures that the mark scheme covers the students' responses to questions and that every examiner understands and applies it in the same correct way. As preparation for the standardisation meeting each examiner analyses a number of students' scripts: alternative answers not already covered by the mark scheme are discussed at the meeting and legislated for. If, after this meeting, examiners encounter unusual answers which have not been discussed at the meeting they are required to refer these to the Principal Examiner.

It must be stressed that a mark scheme is a working document, in many cases further developed and expanded on the basis of students' reactions to a particular paper. Assumptions about future mark schemes on the basis of one year's document should be avoided; whilst the guiding principles of assessment remain constant, details will change, depending on the content of a particular examination paper.

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Marking Guidance for Examiners

GCSE Science Papers

1. General

The mark scheme for each question shows:

- the marks available for each part of the question
- the total marks available for the question
- the typical answer or answers which are expected
- extra information to help the Examiner make his or her judgement and help to delineate what is acceptable or not worthy of credit or, in discursive answers, to give an overview of the area in which a mark or marks may be awarded.

The extra information is aligned to the appropriate answer in the left-hand part of the mark scheme and should only be applied to that item in the mark scheme.

At the beginning of a part of a question a reminder may be given, for example:

where consequential marking needs to be considered in a calculation;

or the answer may be on the diagram or at a different place on the script.

In general the right hand side of the mark scheme is there to provide those extra details which confuse the main part of the mark scheme yet may be helpful in ensuring that marking is straightforward and consistent.

2. Emboldening

- 2.1** In a list of acceptable answers where more than one mark is available 'any **two** from' is used, with the number of marks emboldened. Each of the following lines is a potential mark.
- 2.2** A bold **and** is used to indicate that both parts of the answer are required to award the mark.
- 2.3** Alternative answers acceptable for a mark are indicated by the use of **or**. (Different terms in the mark scheme are shown by a / ; eg allow smooth / free movement.)

3. Marking points

3.1 Marking of lists

This applies to questions requiring a set number of responses, but for which students have provided extra responses. The general principle to be followed in such a situation is that 'right + wrong = wrong'.

Each error/contradiction negates each correct response. So, if the number of error/contradictions equals or exceeds the number of marks available for the question, no marks can be awarded.

However, responses considered to be neutral (indicated as * in example 1) are not penalised.

Example 1: What is the pH of an acidic solution? (1 mark)

Student	Response	Marks awarded
1	4,8	0
2	green, 5	0
3	red*, 5	1
4	red*, 8	0

Example 2: Name two planets in the solar system. (2 marks)

Student	Response	Marks awarded
1	Pluto, Mars, Moon	1
2	Pluto, Sun, Mars, Moon	0

3.2 Use of chemical symbols / formulae

If a student writes a chemical symbol / formula instead of a required chemical name, full credit can be given if the symbol / formula is correct and if, in the context of the question, such action is appropriate.

3.3 Marking procedure for calculations

Full marks can be given for a correct numerical answer, as shown in the column 'answers', without any working shown.

However if the answer is incorrect, mark(s) can be gained by correct substitution / working and this is shown in the 'extra information' column;

3.4 Interpretation of 'it'

Answers using the word 'it' should be given credit only if it is clear that the 'it' refers to the correct subject.

3.5 Errors carried forward

Any error in the answers to a structured question should be penalised once only.

Papers should be constructed in such a way that the number of times errors can be carried forward are kept to a minimum. Allowances for errors carried forward are most likely to be restricted to calculation questions and should be shown by the abbreviation e.c.f. in the marking scheme.

3.6 Phonetic spelling

The phonetic spelling of correct scientific terminology should be credited **unless** there is a possible confusion with another technical term.

3.7 Brackets

(....) are used to indicate information which is not essential for the mark to be awarded but is included to help the examiner identify the sense of the answer required.

PHY2F**Question 1**

question	answers	extra information	mark
1(a) (E)	1800 (N)	allow 1 mark for correct substitution ie 180×10 provided no further steps shown	2
1(b) (E)	3780 or their (a) $\times$ 2.1 correctly calculated joule	allow 1 mark for correct substitution ie 1800 or their (a) $\times$ 2.1 provided no further steps shown accept J accept any clear indication of correct answer	2 1
1(c) (E)	0 work is only done when a force makes an object move	reason does not score if 0 not chosen accept distance moved is zero accept no energy transfer (to the bar) accept the bar is not moving/is stationary 'it' refers to the bar/weights	1 1
Total			7

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Question 2

question	answers	extra information	mark
2(a) (E)	A it is alternating / a.c. or it changes direction/p.d.	only scores if A chosen	1
		accept because B and C are d.c. accept voltage for p.d. it goes up and down is insufficient it is constantly changing is insufficient an answer B and/or C with the reason because it is <u>direct current</u> / <u>d.c</u> scores 1 mark	1
2(b) (E)	too much current (through socket) wiring / socket gets hot	accept electricity for current accept too much power accept socket/circuit overloaded do not accept voltage/p.d for current	1
		accept melts for gets hot accept risk of fire risk of fire in appliances is insufficient ignore reference to sparking overloaded plugs and plugs getting hot or fuses melting is insufficient	1
Total			4

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Question 3

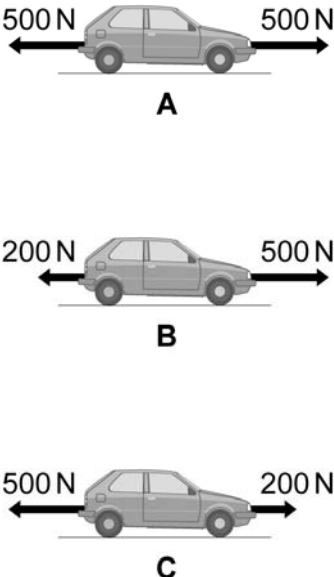
question	answers	extra information	mark
3(a)(i) (E)	bowl the ball faster	accept increase its speed accept a stated speed above 20 m/s increase momentum is insufficient bowl ball with greater power/force is insufficient bowl ball harder is insufficient do not accept increase mass of ball	1
3(a)(ii) (E)	3.2 kg m/s	allow 1 mark for correct substitution i.e. 0.16×20 provided no further steps shown accept any clear indication of correct answer	2 1
3(b)(i) (E)	work done by ball to move stumps	accept transformed into heat / sound accept transferred to surroundings accept transferred to the stumps do not accept absorbed by the stumps	1
3(b)(ii) (E)	the stumps gain momentum equal to momentum lost by ball	accept momentum is <u>conserved</u>	1 1
Total			7

PHY2F**Question 4**

question	answers	extra information	mark
4(a)(i) (E)	24	allow 1 mark for converting time to 600 seconds or showing method ie $14400/10$ or <u>14400</u> 10×60 provided no further steps shown	2
4(a)(ii) (E)	24 or their (a)(i)	ignore any unit	1
4(b)(i) (G)	20 45	both required – either order	1
4(b)(ii) (A)	the block transfers energy to the surroundings		1
Total			5

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Question 5

question	answers	extra information	mark
5(a) (G)	 Diagram A: 500 N ← → 500 N Diagram B: 200 N ← → 500 N Diagram C: 500 N ← → 200 N	3 lines drawn all correct allow 1 mark for each correct line if two or more lines are drawn from any diagram then all these lines are incorrect stationary constant speed slowing down accelerating forwards	3
5(b)(i) (G)	horizontal arrow to the right	judge by eye accept an arrow drawn outside the box if it is labelled correctly	1
5(b)(ii) (G)	horizontal arrow to the left	judge by eye accept an arrow drawn outside the box if it is labelled correctly	1

Question 5 continues on the next page . . .

PHY2F**Question 5 continued . . .**

question	answers	extra information	mark
5(b)(iii) (A)	equal to		1
5(b)(iv) (A)	to measure the forces exerted on the dummy during the impact		1
Total			7

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Question 6

question	answers	extra information	mark
6(a)(i) (E)	friction between the beads and pipe	accept beads rub against the pipe	1
	(cause) <u>electrons</u> to transfer	accept electrons are lost/gained do not accept negatively charged atoms for electrons	1
	from the pipe or to the beads	3 rd mark point only scores if 2 nd mark scores do not accept from the (negatively) charged pipe do not accept to the (positively) charged beads accept negative charge transfer to the beads for 1 mark provided 2 nd or 3 rd marking point not awarded mention of positive charge transfer negates last 2 marking points	1
6(a)(ii) (E)	<u>volume</u> of beads or <u>length</u> of pipe or speed the beads are poured or angle of pipe	accept (75)cm ³ accept use the same pipe poured the same way is insufficient	1
6(b)(i) (E)	the larger the beads the less charge	do not accept inversely proportional negative correlation is insufficient	1

Question 6 continues on the next page . . .

PHY2F**Question 7**

question	answers	extra information	mark
7(a)(i) (E)	(nuclear) fission is the splitting of a (large atomic) nucleus	do not accept particle/atom for nucleus	1
	(nuclear) fusion is the joining of (two atomic) nuclei (to form a larger one)	do not accept particles/atoms for nuclei	1
7(a)(ii) (E)	energy	accept heat/radiation/nuclear energy accept gamma (radiation) do not accept neutrons/neutrinos	1
7(b)(i) (E)	uranium (-235)	accept U (-235) ignore any numbers given with uranium accept thorium accept MOX (mixed oxide) do not accept hydrogen	1
7(b)(ii) (E)	(same) number of protons	accept (same) atomic number accept (same) <u>positive</u> charge ignore reference to number of electrons	1
Total			5

UMS Conversion Calculator
<http://web.aqa.org.uk/UMS/index.php>